## Using microorganisms to harvest electricity -Towards a self-regenerating enzymatic cathode

S. Sané<sup>1</sup>, S. Rubenwolf<sup>1</sup>, R. Zengerle<sup>1,2</sup>, C. Jolivalt<sup>3</sup>, S. Kerzenmacher<sup>1</sup>

<sup>1</sup>University of Freiburg – IMTEK, Department of Microsystems Engineering, Laboratory for

MEMS Applications, Freiburg, Germany

<sup>2</sup>BIOSS Centre for Biological Signalling Studies, Albert-Ludwigs University, Freiburg, Germany <sup>3</sup>Ecole Nationale Supérieure de Chimie de Paris/ Chimie ParisTech

Laboratoire Charles Friedel, Paris, France

Sabine.Sane@imtek.de

The advantage of microbial biofuel cells (mBFCs) is that microorganisms continuously reproduce and thus lifetimes of up to five years have already been reported [1]. While these cells have a lower power density (e.g. 188  $\mu$ W/cm<sup>2</sup> [2]) than enzymatic biofuel cells (eBFCs; e.g. 1450  $\mu$ W/cm<sup>2</sup> [3]), enzyme degradation limits the lifetime of eBFCs to only a few weeks and therefore hinders their long-term use [4].

The aim of our research is to improve the long-term stability of efficient, but currently short-lived enzymatically–catalyzed fuel cell electrodes [4]. We aim to continuously supply catalytically-active enzymes at the electrode using living microorganisms that grow in an electrode-integrated micro-bioreactor. In this way, the advantages of eBFCs and mBFCs would be combined by establishing a fully self-regenerating eBFC.

In the present work, we demonstrate the feasibility of using the crude culture supernatant of the fungus *Trametes versicolor*, which produces laccase as an extracellular enzyme, to supply a biofuel cell cathode. During culturing, aliquots of the supernatant were regularly removed and replaced by fresh medium. The aliquots, containing approximately 2.2 U/mL laccase activity, were transferred to the cathode compartment of a BFC. To record the loadcurve, the current was incrementally increased (steps of  $5.6 \,\mu\text{A/cm}^2\text{*h}$ ) and the cathode potential was measured against a reference electrode (SCE). This resulted in a current density of 74  $\mu$ A/cm<sup>2</sup> at a cathode potential of 0.4 V vs. SCE. The same activity of commercial laccase (Sigma) in SC

medium produced a current density of  $78 \,\mu\text{A/cm}^2$  at the same cathode potential (Figure 1). Hence, we could show that it is possible to use the crude culture supernatant of *T. versicolor* at the electrode, and that the current density obtained is comparable to the current density recorded with commercial laccase. Our work is thus a first step towards the aim of constructing a self-regenerating enzymatic biofuel cell.

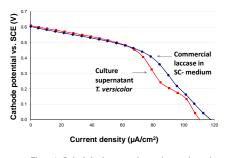


Figure 1: Cathode loadcurves under continuous air purging.

- [1] B.H. Kim et al., Biotechnol. Lett., (2003), pp. 541-545.
- [2] K.P. Nevin et al., Environ. Microbiol., (2008), pp. 2505-2514.
- [3] H. Sakai et al., Energy Environ. Sci., (2009), pp. 133-138.
- [4] S. Rubenwolf et al., Appl. Microbiol. Biotechnol., (2011), pp. 1315-1322.